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TABLE OF CONTENTS



I	Papers published in journals: Academic year 2017-18	Page no
1.	Dr. Vikas nathan	4
2.	Dr. Vikas nathan	5
3.	Dr. Kamayani	6
4.	Dr. Kamayani	6
5.	Dr. Kamayani	6
6.	Dr. Poonam	7
7.	Dr. Poonam	7
8.	Dr. Anjana sharma	8
9.	Dr. Girish sharma	9
II	Papers published in journals: Academic year 2018-19	
10.	Dr. Vikas Nathan	11
11.	Dr. Vikas Nathan	12
12.	Dr. Kirti Singha	13
13.	Dr. Kirti Singha	14
14.	Dr. Kirti Singha	15
15.	Dr. Monika Chandel	16
16.	Dr. Monika Chandel	17
17.	Dr. Monika Chandel	18
18.	Dr. Poonam Chandel	19
19.	Dr. Pawan Kumar	20-21
20.	Dr. Sandeep Chauhan	22

21.	Dr. Kamayani	23
22.	Dr. Kamayani	23
23.	Dr. Kamayani	23
III	Papers published in journals: Academic year 2019-20	
24.	Dr. Vikas Nathan	25
25.	Dr. Vikas Nathan	26
26.	Dr. Kirti Singha	27
27.	Dr. Kirti Singha	28
28.	Dr. Monika Chandel	29
29.	Dr. Monika Chandel	30
IV	Papers published in journals: Academic year 2020-21	
30.	Dr. Kirti Singha	32
31.	Dr. Monika Chandel	33
32.	Dr. Anjana Sharma	34
V	Papers published in journals: Academic year 2021-22	
33.	Dr. Rajesh Kumar Dhorta	36
34.	Dr. Rajesh Kumar Dhorta	37
35.	Dr. Rajesh Kumar Dhorta	37
36.	Dr. Satya Narain Snehi	38
37.	Dr. Satya Narain Snehi	39
38.	Dr. Satya Narain Snehi	40
39.	Dr Aditya Dulta	41
40.	Dr Aditya Dulta	42



Papers published in journals: Academic year 2017-18




Molecular interaction studies of antibiotic drug Doxycycline Hyclate with aqueous mannitol using volumetric and acoustic methods

Shashi Kant Sharma  , Vikas Nathan, Dinesh Kumar, Kamal Kishore

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Highlights



- Molecular interactions between Doxycycline Hyclate and mannitol are studied.
- The volumetric and acoustic methods are used to interpret results.
- Temperature variation of parameters studied
- Solute-solvent interactions are present.
- Doxycycline Hyclate behaves as a structure maker in water-mannitol system.


Abstract



The behaviour of Doxycycline Hyclate (DH) in aqueous mannitol solution was studied to explore molecular interactions at different temperatures. The volumetric and acoustic studies were used for investigating the interactions of drug Doxycycline Hyclate in water and aqueous mannitol system. The Density (ρ) and ultrasonic velocity (u) of Doxycycline Hyclate in water and in (0.1, 0.2 and 0.4) mol·kg⁻¹ aqueous solutions of mannitol have been measured at (305.15, 310.15 and 315.15K) temperatures and atmospheric pressure. The density data was analysed with the help of Masson's equation. The positive value of (Φ_v^0) for DH indicates solute-solvent interactions. The solute-solute interactions were determined from Masson's coefficient, (S_v) in water-mannitol system at different temperatures. The ultrasonic velocity data of DH in water and water-mannitol system were used to determine adiabatic compressibility (β), intermolecular free length (L_f), and specific acoustic impedance (Z). The structure making/breaking behaviour of DH in water



Molecular interaction studies of L-alanine and L-phenylalanine in water and in aqueous citric acid at different temperatures using volumetric, viscosity and ultrasonic methods

Dinesh Kumar, Shashi Kant Lomesh  , Vikas Nathan

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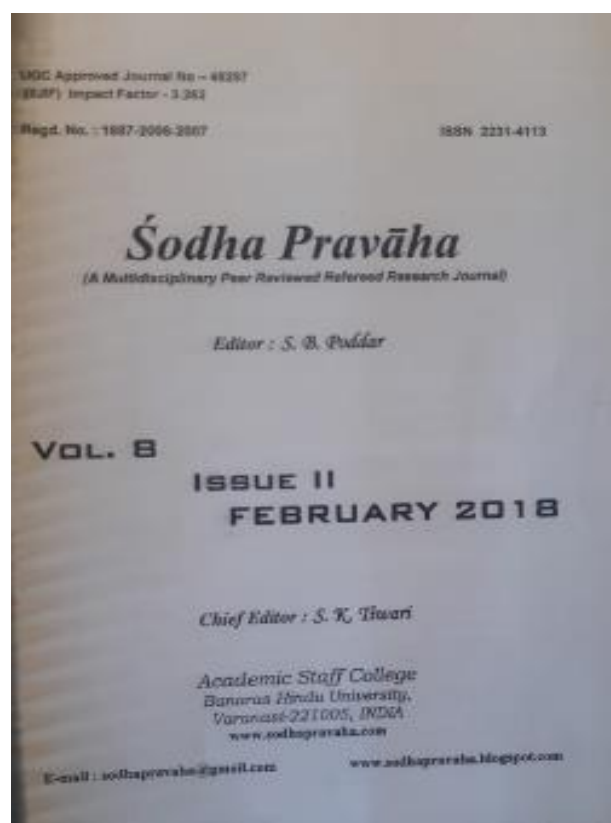
Highlights

- Molecular interactions between L-alanine/L-phenylalanine and aqueous citric acid are studied.
- Temperature variation of partial molar volume, Jone-Dole's coefficient B, adiabatic compressibility are studied.
- Solute-solvent interactions predominate over solute-solute interactions.
- L-Alanine behaves as structure breaker whereas L-phenylalanine behaves as a structure maker in water-citric acid system.

Abstract

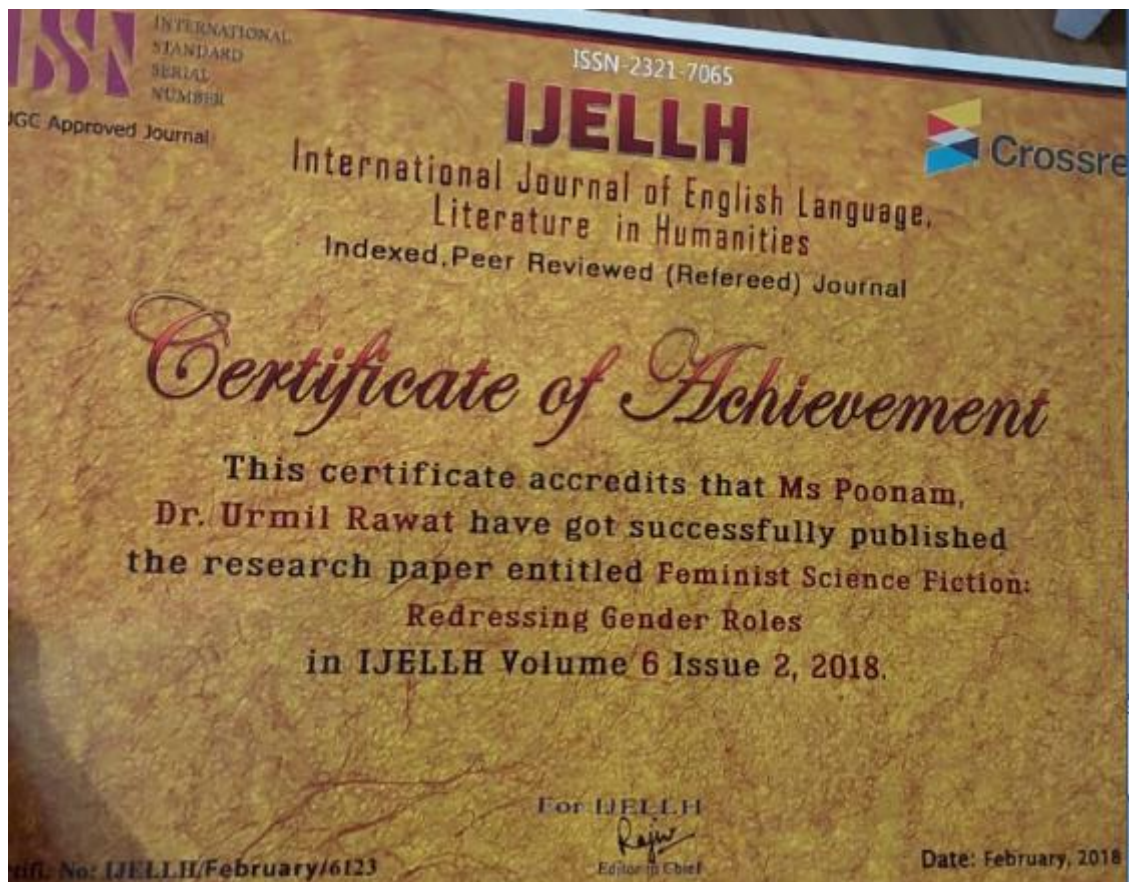
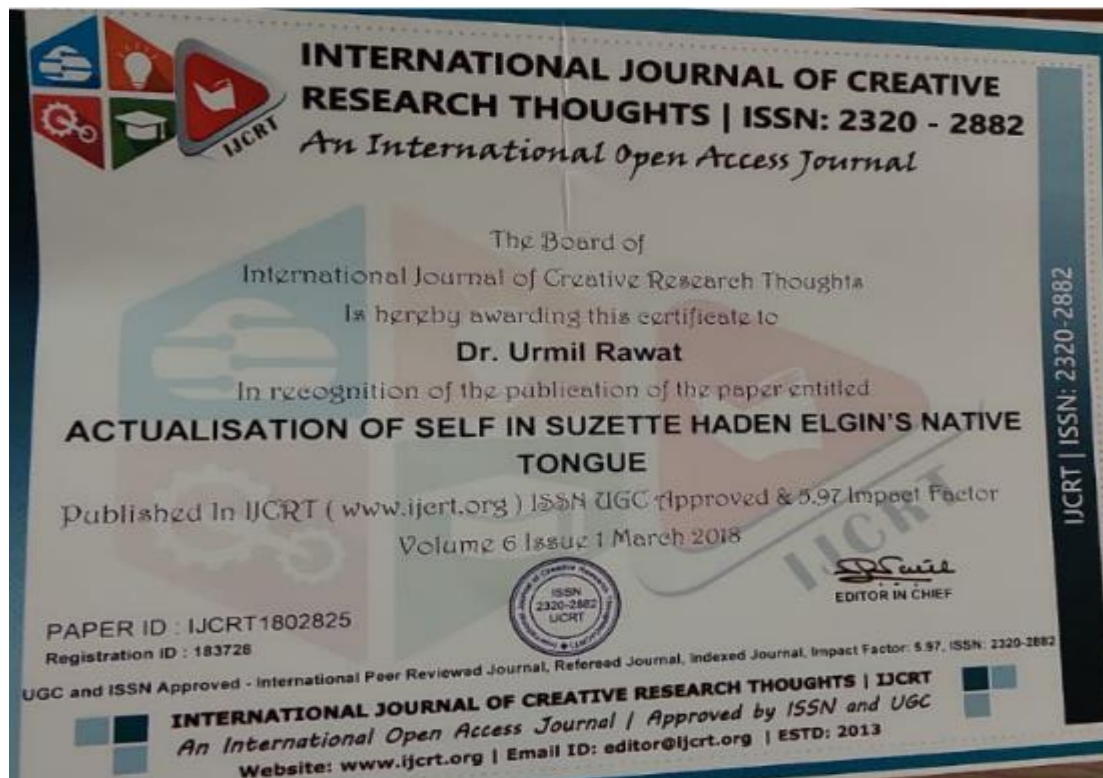
Densities, ultrasonic speeds and viscosities of L-alanine and L-phenylalanine in water and in 0.1 mol·kg⁻¹ aqueous citric acid solutions were measured over the temperature range (298.15 to 313.15)K with interval of 5K at atmospheric pressure. From these experimental data apparent molar volume, limiting apparent molar volume and the slope, partial molar expansibilities, adiabatic compressibility, transfer volume, Falkenhagen coefficient, Jone-Dole's coefficient, the temperature derivative of Jone-Dole's coefficient, intermolecular free length, specific acoustic impedance, and molar compressibility were calculated. The results are interpreted in terms of solute – solute and solute-solvent interactions in these systems. It has also been observed that L-alanine acts as a structure breaker whereas L-phenylalanine acts as a structure maker in aqueous citric acid.

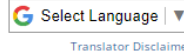
Dr. kamayani



<i>Sodha Pravaha</i> , ISSN 2231-4113, Vol. 8, Issue II, February 2018	
Copper(II) Complexes of Schiff Base Ligand Derived From 3-Methoxyvalerylaldehyde With Semicarbazide and Thiosemicarbazid-Preparation and Biological Testing with Vastation Sanjeev Kumar, Department of Chemistry, Jai Prakash University, Chapra	131-140
Importance of Iron in Development Niraj Kumar, Research Scholar (Chemistry), L.N.M.U., Darbhanga	141-143
A Study on Interventions/Provisions for Safe Guarding Child Labour Dr. Ashwini Sahaya, +2 Teacher (Faculty of Commerce), D.A.V. Govt. Inter College, Noida, (Uttar) India	144-150
The Action of Extremely Weak Alternating Magnetic Field at Cyclotron Frequency on Aqueous Solution of Amino Acid Sudhakar Pawley, Assistant Professor, Department of Physics, Govt. P.G. College, Magaraha, Mirzapur - 231304 (India)	151-154
Women's Economic Independence and the Probability of Divorce: A Review and Re-examination Swati, Research Scholar, Banaras Hindu University Varanasi UP 221005	155-160
Children Rights with Reference to Various Provisions of the Constitution Pravara Kumar, Research Scholar, Dept. of P.A.L.R, Panna University, Panna	161-166
Social Development and Justice Among Dalits Shashi Kishore, U.G.C. NET, Research Scholar, Dept. of Sociology, Panna University, Panna	167-169
Information & Communication Technology : A Conceptual Framework Arjun Prasad Kumar, Research Scholar, Faculty of Management, B.B.A.B.U., Meerut/Uttar	170-174
Education and Skill Development: A Conceptual Framework Nandini Kumar Sak, Research Scholar, Faculty of Management, B.B.A.B.U., Meerut/Uttar	175-178
Class Mobility in Contemporary Spoken English Kamayani Bhatt, Department of English, Govt. College, Thang, Sharda, UP	179-182
Lohia as a Theorist of Seven Revolutions Jay Krishna Kumar, Department of Pol. Science, B.N.M.U., Madhupur, Bihar	183-188
The Livelihood Conditions of Handloom Weavers of Varanasi Feroz Khatun, Research Scholar, Department of Economics, P.S.S, Banaras Hindu University, Varanasi, 221005	189-195
Effect of Different Levels of Meloidogyne incognita and Nematodes on Plant Growth of Chickpea and Population Dynamics of Nematode K. K. Singh, Department of Plant Pathology, College of Agriculture, Banars University of Agriculture Science and Technology, Banars-221001 (U.P.) K.K.Singh, SRVM, Varanasi-221005 (U.P.)	196-200

Dr. Poonam Verma





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2019

Means, moments and Newton's inequalities

R. Sharma, A. Sharma, R. Saini, G. Kapoor

Rocky Mountain J. Math. 49(5): 1667-1677 (2019). DOI: 10.1216/RMJ-2019-49-5-1667

- ABOUT
- FIRST PAGE
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Abstract

It is shown that Newton's inequalities and the related Maclaurin's inequalities provide several refinements of the fundamental arithmetic-geometric-harmonic mean inequality and Sierpinski's inequality in terms of the means and variance of positive real numbers. We also obtain some inequalities involving third and fourth central moments of real numbers.

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ROCKY MOUNTAIN
JOURNAL OF MATHEMATICS
Volume 49, Number 5, 2019

MEANS, MOMENTS AND NEWTON'S INEQUALITIES

R. SHARMA, A. SHARMA, R. SAINI AND G. KAPOOR

ABSTRACT. It is shown that Newton's inequalities and the related Maclaurin's inequalities provide several refinements of the fundamental arithmetic-geometric-harmonic mean inequality and Sierpinski's inequality in terms of the means and variance of positive real numbers. We also obtain some inequalities involving third and fourth central moments of real numbers.

1. Introduction. Let x_1, x_2, \dots, x_n denote n real numbers. Their r th moment and r th central moments are, respectively, the numbers

$$(1.1) \quad m'_r = \frac{1}{n} \sum_{i=1}^n x_i^r$$

and

$$(1.2) \quad m_r = \frac{1}{n} \sum_{i=1}^n (x_i - m'_1)^r,$$

where $r = 1, 2, \dots$. Note that $m_1 = 0$, m'_1 is the arithmetic mean and m_2 is the variance of n real numbers x_1, x_2, \dots, x_n . It is customary to denote m'_1 and m_2 , respectively, by A and s^2 . The geometric mean (G) and harmonic mean (H) of n positive real numbers x_1, x_2, \dots, x_n are, respectively, the numbers

$$G = \left(\prod_{i=1}^n x_i \right)^{1/n} \quad \text{and} \quad H = \left(\frac{1}{n} \sum_{i=1}^n \frac{1}{x_i} \right)^{-1}.$$

The k th elementary symmetric function C_k of x_1, x_2, \dots, x_n is the sum of the products taken k at a time of different x_i 's. The k th elementary

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Keywords and phrases. Arithmetic mean, geometric mean, moments, Newton's identities.

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2019

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1. Introduction. Let x_1, x_2, \dots, x_n denote n real numbers. Their r th moment and r th central moments are, respectively, the numbers

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where $r = 1, 2, \dots$. Note that $m_1 = 0$, m'_1 is the arithmetic mean and m_2 is the variance of n real numbers x_1, x_2, \dots, x_n . It is customary to denote m'_1 and m_2 , respectively, by A and s^2 . The geometric mean (G) and harmonic mean (H) of n positive real numbers x_1, x_2, \dots, x_n are, respectively, the numbers

$$G = \left(\prod_{i=1}^n x_i \right)^{1/n} \quad \text{and} \quad H = \left(\frac{1}{n} \sum_{i=1}^n \frac{1}{x_i} \right)^{-1}.$$

The k th elementary symmetric function C_k of x_1, x_2, \dots, x_n is the sum of the products taken k at a time of different x_i 's. The k th elementary


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Physicochemical approach to study the solute-solute and solute-solvent interactions of drug Levofloxacin hemihydrate in aqueous sorbitol solutions at different temperatures: Volumetric, acoustic and conductance studies

[Shashi Kant Lomesh](#)  , [Madhu Bala](#), [Vikas Nathan](#)

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Highlights

- Different possible interactions between LFH in water and aqueous sorbitol systems are studied.
- The various possible interactions are interpreted using volumetric, acoustic and conductance studies.
- Solute-solvent interactions predominate over solute-solute interactions.
- All these studies confirmed the structure maker behaviour of LFH in water and water-sorbitol system.

Abstract

The interactions of third generation fluoroquinolone antibiotic drug Levofloxacin hemihydrate (LFH) with the naturally occurring polyol i.e. sorbitol as a function of temperature have been investigated by volumetric, acoustic and conductance methods. Densities, speeds of sound and conductance values of LFH ($0.001-0.01 \text{ mol}\cdot\text{kg}^{-1}$) in water and aqueous sorbitol ($0.002, 0.004$ and $0.006 \text{ mol}\cdot\text{kg}^{-1}$) have been measured at different temperatures ($300.15, 305.15, 310.15$ and 315.15 K). The experimental density data was used to calculate apparent molar volume (Φ_V), limiting apparent molar volume (Φ_V^0), Masson's coefficient (S_V), partial molar expansibilities (Φ_E^0), transfer volume $\Delta_{tr}\Phi_V^0$ and



Volumetric and acoustic methods for investigating molecular interactions of antibiotic drug doxycycline hyclate in water and in aqueous solution of sodium chloride and potassium chloride at different temperatures (293.15–313.15) K

Shashi Kant Lomesh, Vikas Nathan *, Madhu Bala, Poonam Thakur

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ABSTRACT

The solution properties of antibiotic drug Doxycycline Hyclate (DH) were investigated to explore molecular interactions with aqueous NaCl and KCl solutions around their physiological cell concentrations at different temperatures. Apparent molar properties i.e. apparent molar volume (ϕ_V) and apparent molar isentropic compression (ϕ_K) for DH within the concentration range (0.0020–0.0140) mol·kg⁻¹ in water and in (0.005, 0.012 and 0.140) mol·kg⁻¹ aqueous NaCl and KCl solutions were calculated from experimentally measured densities (ρ) and Ultrasonic speed (u) values at $T = (293.15, 298.15, 303.15, 308.15$ and $313.15)$ K temperatures and at $P = 0.1$ MPa pressure. The data of apparent molar properties were utilised to obtain various derived parameters such as limiting apparent molar volume, (ϕ_V^0), limiting apparent molar isentropic compression, (ϕ_K^0), limiting apparent molar volume of transfer, $\Delta\phi_V^0$, limiting apparent molar isentropic compression of transfer, $\Delta\phi_K^0$, limiting apparent molar expansibility, ϕ_E^0 , and thermal expansion coefficient, α . The structure making/breaking ability of DH in aqueous NaCl and KCl were determined on the basis of Hepler's Equation i.e. on the basis of sign of $(\frac{d^2\phi_V^0}{dT^2})_P$. The results obtained from all these thermodynamic parameters were discussed in terms of solute-solute and solute-solvent interactions in these systems. The volumetric, compressibility and acoustic data suggests the existence of strong solute-solvent interactions in the studied systems.

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1. Introduction

Drug as defined in pharmacology, a chemical substance used in the treatment, cure, prevention or diagnosis of disease or used to otherwise enhance physical or mental well-being [1,2]. Physicochemical properties of drugs are important to understand the drug action at the molecular level [3,4]. The binding tendencies of biologically important molecules in the presence of various solvents can be interpreted by studying their physicochemical properties [4–6]. Interaction of drugs with physiologically important molecules in the living organism i.e. in blood, membranes, intra- and extra-cellular fluids are of utmost significance to understand the pharmacodynamics and pharmacokinetics of drugs [7,8]. These interactions may include ionic, covalent, charge transfer, hydrogen bonding, ion-dipole interactions, or hydrophobic hydration [9,10]. Several thermodynamic and physicochemical measurements for drugs in aqueous solutions and in liquid mixtures have been carried

out to understand the nature of molecular interactions existing in these systems [11,12]. The study of aqueous phase chemistry of drugs in the presence of biologically important molecules including proteins, amino acids, peptides, electrolytes, carbohydrates, metal ions, colloidal micelles [13] and bionanomaterials as drug delivery tools [14] etc. by thermodynamic and volumetric methods constitute an important topic of research due to their important function in the fields related to biology, medicine, target drug delivery, cosmetics etc. [15].

The Tetracyclines referred to as broad-spectrum antibiotics, due to their ability to fight many different types of bacteria (Gram-positive and Gram-negative bacteria), made up of four fused rings are one of the most important class of drugs, as they are not only used as antibiotic agents but also have novel multiple therapeutic potential [16,17]. DH, (see Fig. 1(a) and (b)) a non Newtonian, synthetic broad spectrum antibiotic used for the treatment of many bacterial infections, such as urinary tract infections, acne, gonorrhoea, Anthrax, Lyme disease, Plague, *P. falciparum* (malaria) and periodontitis is an active tetracycline class medicament that shows powerful effect against infections caused by several micro-organisms groups such as *Haemophilus ducreyi*, *Yersinia pestis*, *Legionella*, *Chlamydia*, *Mycoplasma* and *Borrelia* [18,19].

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Influence of Ho–Ni–Mn substitution on the structural and magnetic behavior of Ba–Sr Co₂Z-type nanohexaferrites extension up to Mossbauer investigations

Kirti Singha¹ · Virender Pratap Singh^{1,2} · Monika Chandel¹ · Nain Jeet Singh Negi³ · Susheel Kalia⁴ · R. K. Kotnala⁵Received: 2 August 2019 / Accepted: 23 October 2019
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Abstract

A series of Co₂Z-type Ba–Sr nanohexaferrites Ba_{1.5}Sr_{1.5}Co_{2-z}Ho_zMn_xNi_yFe_{24-x-y}O₄₁ (z = 0.0, 0.05, 0.10, 0.15, 0.20, x = y = 0.0, 0.25, 0.5, 0.75, 1.00) have been synthesized using sol–gel auto-combustion synthesis route. The effect of Ho–Ni–Mn substitutions on crystallographic and magnetic properties of synthesized nanohexaferrites was investigated using XRD, VSM, and Mössbauer spectroscopy. Microstructural analysis showed single-phase crystal structures without any impurities and hexagonal with the space group *P63/mmc*. The average variation in crystallite size ranges from 43 to 60 nm with a slight increase in X-ray density and appreciable decrease in porosity was observed for different dopants. FE-SEM (Nova Nano SEM-450) substantiates the hexagonal structure and HR-TEM images assisted with SAED pattern confirm the crystalline quality and FWHM of the material, which significantly support the XRD results. FTIR spectra showed two characteristic metal stretching peaks in the range of 400–600 cm⁻¹ due to the substitution of Ho–Ni–Mn. Magnetic measurements show maximum magnetic saturation (*M_s*) at 44.04 emu g⁻¹ and elevated value of coercivity (*H_c*) 2240e imparting typical characteristics of soft ferrite with high coercivity. Mössbauer analysis with least squares fit sextets of six distinguishable sites at room temperature for all samples substantially supports the results of VSM. The materials with large coercivity are useful in permanent magnet applications. The prepared composites could be useful for applications in microwave absorbing materials, magnetic storage, and the miniaturization of antennas for wireless communication devices.

1 Introduction

Hexaferrites are magnetic iron oxides possessing hexagonal structure. The magnetic behavior of these materials is considerably related to their crystal structures and is ferrimagnetic. Magnetocrystalline anisotropy (MCA) is one of the important characteristics of these materials initiated from

the induced magnetization and has a preferred orientation inside the crystal structure. The Z-type ferrites belong to the families of hexagonal ferrites, generally denoted as Co₂Z and have chemical composition Ba₃Me₂Fe₂₄O₄₁, where Me is a divalent transition metal ion. These ferrites have planar magnetic anisotropy even at room temperature and a much higher magnetic storage capacity than spinel ferrites in the high-frequency region (over 1 GHz). Therefore, the utilization of Ba₃Co₂Z ferrite to produce microwave frequency devices has been reported such as inductors, isolators, and communication devices in the high-frequency region [1]. Z-type hexaferrites are interesting materials for more exciting applications, because of their dynamic structural and magnetic properties, besides their scope as magnetic recording, data storage materials, and high-frequency applications. Several research studies have been reported to improve the magnetic properties of Ba hexaferrite by substituting the various ions such as Zn²⁺, Ni²⁺, La³⁺, and Al³⁺ in Co²⁺ and Fe³⁺ in the crystal structure of hexaferrite [2–4]. Moreover, elemental doping is an efficient way to modify the magnetocrystalline structures of Co₂Z-type hexaferrites and therefore changes

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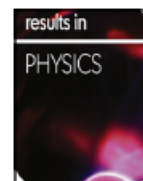
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Analysis of Cd²⁺ and In³⁺ ions doping on microstructure, optical, magnetic and mössbauer spectral properties of sol-gel synthesized BaM hexagonal ferrite based nanomaterials

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ABSTRACT

Precise studies of cadmium and indium doped barium hexagonal ferrites having chemical composition Ba_{0.7}Nd_{0.3}Cd_{x/2}In_{x/2}Fe_{1.2-x}O₁₉ (x = 0.0, 0.1, 0.2, 0.3) have been performed by sol-gel auto-combustion method in which ethylene glycol was used as a gel precursor. The structural, morphological, optical, elemental and magnetic properties have been studied by using various techniques like XRD, FESEM, FTIR, EDS and VSM. The XRD patterns shows characteristic (110), (008), (107), (114), (108), (203), (205), (206), (1011), (300), (217), (2011), (220), (2014) peaks along with the presence of secondary phase confirming the formation of hexagonal structure with an average crystallite size of 43–59 nm. FESEM supports the formation of hexagonal, dense and agglomerated nanoparticles. The Vibronic study using infrared radiation was carried by FTIR analysis reveal the various configuration modes with hexagonal symmetry of prepared nanoparticles. The magnetic measurements have been studied at room temperature indicates that saturation magnetization (M_s) and magnetic moment (n_B) found to be of range 40–86 emu/g and 7.97–17.23 μ_B. The precise magnetic studies made it possible to reveal that

saturation magnetization (M_s) increases with the cadmium and indium concentration for x = 0.1 and after that it decreases for x = 0.2, 0.3 which may be due to the difference in the magnetic moments of Cd, In and Fe ions. Due to high value of saturation magnetization (M_s), it can be used for applications in the field of high density recording storage devices and also, this magnetic change has been explained on the basis of exchange interactions. The room temperature Mössbauer spectra of all the nano-sized materials shows normal Zeeman splitting consisting of six merged line patterns which indicates the formation of ferromagnetic phase that supports the magnetic properties.

Introduction

An iron ore based on iron (III) oxide known as magnetite (Fe₂O₃), called ferrites with combination of divalent cation Me having composition MeFe₂O₄. Each synthesized sample has its own advantages and disadvantage on the basis of various parameters related to different properties of prepared samples [1]. The optimum parameters like crystalline size, grain size, configuration modes, resistivity, dielectric constant, dielectric loss, magnetic loss factor, saturation magnetization, coercivity, remanent magnetization, magnetic moment etc. will depend upon the sintering temperature [2], type of dopants and preparation technique [3] for synthesizing samples. The ferrites are classified into

two categories on the basis of chemistry of magnetism as hard and soft ferrites. On the basis of their structure, ferrites are classified into four categories as spinel, gamet, ortho and hexagonal. The compounds with hexagonal crystal structure are called hexagonal ferrites or hexaferrites. Hexagonal ferrites gain much attention due to its various applications as many tonnes of magnetic materials are produced by using these hexaferrites world widely [4]. All hexaferrites are ferrimagnetic in nature and due to induced magnetization within the crystal structure; they have preferred orientation, which divide them into two groups: First is known as uniaxial hexaferrites having an easy axis of magnetization and second is known as ferroplana or hexaplana ferrites having an easy plane or cone of magnetization. The hexagonal ferrites

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Review

A review on structural, electrical and magnetic properties of Y-type hexaferrites synthesized by different techniques for antenna applications and microwave absorbing characteristic materials

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Abstract: In the present review paper, we have explained the structure of Y-type hexagonal ferrite and various synthesis techniques. This paper also includes structural, electrical, magnetic properties and applications of Y-type hexaferrites and focusses on their use in antenna applications and microwave absorbing characteristic materials. Ferromagnetic nature of hexaferrites cause the induction of magnetisation within the crystal structure, which divide them into two groups: First with easy axis of magnetisation is known as uniaxial hexaferrites and second is known as ferroplana having easy plane of magnetisation. The excellent magnetic properties of Y-type ferrites make them useful in the devices operating at high frequency range. The persistence of high refractive index upto 1 GHz enables these hexagonal ferrites useful in UHF antenna designs with small dimensions.

The doping in Y-type hexaferrites affect all the properties. Current developments in Y-type hexaferrites will be explained in detail in the review of literature related to Y-type hexaferrites for the last 25 years, i.e. from 1994 to 2019 in this review paper.

Keywords: Y-type hexaferrite; structural properties; applications of ferrites; microwave absorbing properties and antenna applications

Abbreviations: μ_r : relative permeability (μ = magnetic permeability); ϵ_r : relative permittivity; ρ : density, in $\text{g}\cdot\text{cm}^{-3}$; B: applied magnetic field (flux density), in SI units of T; dB: decibel, a



Influence of Ho–Ni–Mn substitution on the structural and magnetic behavior of Ba–Sr Co₂Z-type nanohexaferrites extension up to Mossbauer investigations

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Abstract

A series of Co₂Z-type Ba–Sr nanohexaferrites Ba_{1.5}Sr_{1.5}Co_{2-z}Ho_zMn_xNi_yFe_{24-x-y}O₄₁ (z = 0.0, 0.05, 0.10, 0.15, 0.20, x = y = 0.0, 0.25, 0.5, 0.75, 1.00) have been synthesized using sol–gel auto-combustion synthesis route. The effect of Ho–Ni–Mn substitutions on crystallographic and magnetic properties of synthesized nanohexaferrites was investigated using XRD, VSM, and Mössbauer spectroscopy. Microstructural analysis showed single-phase crystal structures without any impurities and hexagonal with the space group *P63/mmc*. The average variation in crystallite size ranges from 43 to 60 nm with a slight increase in X-ray density and appreciable decrease in porosity was observed for different dopants. FE-SEM (Nova Nano SEM-450) substantiates the hexagonal structure and HR-TEM images assisted with SAED pattern confirm the crystalline quality and FWHM of the material, which significantly support the XRD results. FTIR spectra showed two characteristic metal stretching peaks in the range of 400–600 cm⁻¹ due to the substitution of Ho–Ni–Mn. Magnetic measurements show maximum magnetic saturation (*M_s*) at 44.04 emu g⁻¹ and elevated value of coercivity (*H_c*) 2240e imparting typical characteristics of soft ferrite with high coercivity. Mössbauer analysis with least squares fit sextets of six distinguishable sites at room temperature for all samples substantially supports the results of VSM. The materials with large coercivity are useful in permanent magnet applications. The prepared composites could be useful for applications in microwave absorbing materials, magnetic storage, and the miniaturization of antennas for wireless communication devices.

1 Introduction

Hexaferrites are magnetic iron oxides possessing hexagonal structure. The magnetic behavior of these materials is considerably related to their crystal structures and is ferrimagnetic. Magnetocrystalline anisotropy (MCA) is one of the important characteristics of these materials initiated from

the induced magnetization and has a preferred orientation inside the crystal structure. The Z-type ferrites belong to the families of hexagonal ferrites, generally denoted as Co₂Z and have chemical composition Ba₃Me₂Fe₂₄O₄₁, where Me is a divalent transition metal ion. These ferrites have planar magnetic anisotropy even at room temperature and a much higher magnetic storage capacity than spinel ferrites in the high-frequency region (over 1 GHz). Therefore, the utilization of Ba₃Co₂Z ferrite to produce microwave frequency devices has been reported such as inductors, isolators, and communication devices in the high-frequency region [1]. Z-type hexaferrites are interesting materials for more exciting applications, because of their dynamic structural and magnetic properties, besides their scope as magnetic recording, data storage materials, and high-frequency applications. Several research studies have been reported to improve the magnetic properties of Ba hexaferrite by substituting the various ions such as Zn²⁺, Ni²⁺, La³⁺, and Al³⁺ in Co²⁺ and Fe³⁺ in the crystal structure of hexaferrite [2–4]. Moreover, elemental doping is an efficient way to modify the magnetocrystalline structures of Co₂Z-type hexaferrites and therefore changes

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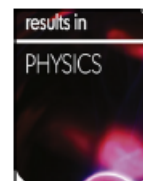
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Analysis of Cd²⁺ and In³⁺ ions doping on microstructure, optical, magnetic and mössbauer spectral properties of sol-gel synthesized BaM hexagonal ferrite based nanomaterials

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ABSTRACT

Precise studies of cadmium and indium doped barium hexagonal ferrites having chemical composition Ba_{0.7}Nd_{0.3}Cd_{x/2}In_{x/2}Fe_{1.2-x}O₁₉ (x = 0.0, 0.1, 0.2, 0.3) have been performed by sol-gel auto-combustion method in which ethylene glycol was used as a gel precursor. The structural, morphological, optical, elemental and magnetic properties have been studied by using various techniques like XRD, FESEM, FTIR, EDS and VSM. The XRD patterns shows characteristic (110), (008), (107), (114), (108), (203), (205), (206), (1011), (300), (217), (2011), (220), (2014) peaks along with the presence of secondary phase confirming the formation of hexagonal structure with an average crystallite size of 43–59 nm. FESEM supports the formation of hexagonal, dense and agglomerated nanoparticles. The Vibronic study using infrared radiation was carried by FTIR analysis reveal the various configuration modes with hexagonal symmetry of prepared nanoparticles. The magnetic measurements have been studied at room temperature indicates that saturation magnetization (M_s) and magnetic moment (n_B) found to be of range 40–86 emu/g and 7.97–17.23 μ_B. The precise magnetic studies made it possible to reveal that

saturation magnetization (M_s) increases with the cadmium and indium concentration for x = 0.1 and after that it decreases for x = 0.2, 0.3 which may be due to the difference in the magnetic moments of Cd, In and Fe ions. Due to high value of saturation magnetization (M_s), it can be used for applications in the field of high density recording storage devices and also, this magnetic change has been explained on the basis of exchange interactions. The room temperature Mössbauer spectra of all the nano-sized materials shows normal Zeeman splitting consisting of six merged line patterns which indicates the formation of ferromagnetic phase that supports the magnetic properties.

Introduction

An iron ore based on iron (III) oxide known as magnetite (Fe₂O₃), called ferrites with combination of divalent cation Me having composition MeFe₂O₄. Each synthesized sample has its own advantages and disadvantage on the basis of various parameters related to different properties of prepared samples [1]. The optimum parameters like crystalline size, grain size, configuration modes, resistivity, dielectric constant, dielectric loss, magnetic loss factor, saturation magnetization, coercivity, remanent magnetization, magnetic moment etc. will depend upon the sintering temperature [2], type of dopants and preparation technique [3] for synthesizing samples. The ferrites are classified into

two categories on the basis of chemistry of magnetism as hard and soft ferrites. On the basis of their structure, ferrites are classified into four categories as spinel, gamet, ortho and hexagonal. The compounds with hexagonal crystal structure are called hexagonal ferrites or hexaferrites. Hexagonal ferrites gain much attention due to its various applications as many tonnes of magnetic materials are produced by using these hexaferrites world widely [4]. All hexaferrites are ferrimagnetic in nature and due to induced magnetization within the crystal structure; they have preferred orientation, which divide them into two groups: First is known as uniaxial hexaferrites having an easy axis of magnetization and second is known as ferroxplana or hexaplana ferrites having an easy plane or cone of magnetization. The hexagonal ferrites

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Review

A review on structural, electrical and magnetic properties of Y-type hexaferrites synthesized by different techniques for antenna applications and microwave absorbing characteristic materials

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Turbulent Kashmir: Unresolved issue of the Partition

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The 200 year long British rule came to an end on August 15, 1947 with the Partition of India into two nations, namely Pakistan and India. It was a violent separation of communities who had lived together for generations. This tragic divide claimed thousands of lives, displaced millions and left behind it the legacy of the disputed Kashmir issue, which is still unresolved. Because of this dispute the two countries are spending huge sums of money on arms and this phenomenon has greatly affected the economic development of both the countries. Insurgency and terrorism fuelled by Pakistan to create instability in the region to pressurize India for Kashmir's merger with Pakistan has its roots in the great divide of 1947. In this context the quest of the paper would be to find out the factors which led to and widened this devastating divide and to explore ways and possible solutions to the Kashmir dispute, for lasting peace in the region and better relations between India and Pakistan.

India and Pakistan have shared a feeling of "mutual distrust" ever since the partition of India. Kashmir has been the bedrock issue between both the nations and has been an unresolved boundary dispute.

This distrust was an outcome of the colonial legacy of "*divide and rule*". The colonial policy of divide and rule pitted the Muslims against the Hindus. The policy gave rise, amongst the Muslims, to a sentiment of "Islam being in danger" in the Hindu dominated Indian society. The growth of Islamic nationalism and hence the demand for a separate Muslim state was an outcome of this paranoia.¹

Since the days of Syed Ahemad Khan, in 1880, a basic theme of the Muslim communities was that if the British left India, the Hindus, because they were a majority, would dominate the Muslims and totally override their interests. Even a cultured and liberal person like Jinnah now catered to the worst feelings of fear and hatred. In his presidential address to the Muslim League in 1938 he said: "The high Command of the Congress is determined, absolutely determined, to crush all other communities and cultures in this country and to establish Hindu Raj".² On the other hand Hindu Communalists did not lag behind. V. D. Savarkar, the then President of Hindu Mahasabha, went to the extent of saying that the Muslims want to reduce the Hindus to the position of helots in their own land.³

The idea that India and Pakistan are two separate communities divided by religion took birth with the aid of the British midwife. For the perpetuation of British rule it was necessary to reduce the influence of the congress and to create rivals to the Hindu community. The obvious strategy to detract the Hindus was to show favors to the Muslims. There are numerous instances of this policy from the earlier times. In the

Dr. Pawan Kumar

First principles studies of Si clusters

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First Principles Studies of Si Clusters

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Abstract. Clusters are in between atoms and bulk matter. Using Quantum Espresso *ab-initio* package we found most stable structures for Si₃, Si₇, Si₁₂, Si₁₃, Si₁₃H₁₂, Si₁₃H₁₄ and Si₁₃H₁₆ and visualized the resulting structures using XCRYSDEN. We also analyzed these structures using symmetry, density of states, valence band width and band gap values. In case of hydrogenated Si clusters broader valence band and narrower band gap has been found well to correlates with stability.

INTRODUCTION

Clusters are the aggregate of atoms or molecules with a well-defined number of constituents ranging from few to several thousands of atoms. They form the intermediate between atoms (or molecules) and the bulk matter and the study of clusters actually study of the evolution of the physical matter from atoms towards bulk solids.

Study of clusters can help us in the potential use them in the technological applications[1-2]. Study of cluster means how the clusters are different from atoms and bulk, their behavior changes with size, their geometry and structure. Further, how variations in their size are related to the electronic, magnetic, optical and chemical properties. These studies can lead us to tailor their properties to get desired properties.

The bulk crystalline matter is arranged in a periodic form while clusters are much smaller and are usually just non-periodic aggregates of atoms. Due to these variations in the properties of clusters and bulk matter the entire theoretical framework used to describe the bulk matter is simply not applicable to clusters.

METHODS

The tool which we have used in our study is Quantum-Espresso package. It is a full *ab initio* package implementing electronic structure and energy calculations. Inside this package, PWSCF is the code which is used to perform total energy calculations. We have used generalized gradient approximation (GGA) of density functional theory using QE. Self-consistent calculations are done after supplying atomic positions and atoms are moved to decrease system energy. This is repeated till energy convergence is not achieved. XCRYSDEN has been used to visualize the resulting structures.

RESULT AND DISCUSSIONS

We have studied properties of Si clusters up to n=13. The properties discussed are structure, energy, density of states and energy level diagrams. We have also hydrogenated the resulting cluster of n=13 with hydrogen atoms.

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A Novel Approach for the Removal of Fluoride from Water Using Active Hydrogels

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ABSTRACT

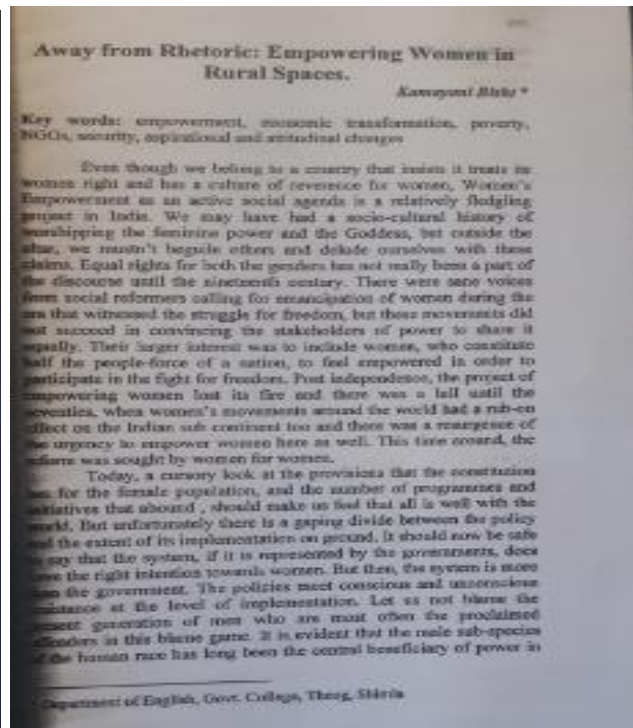
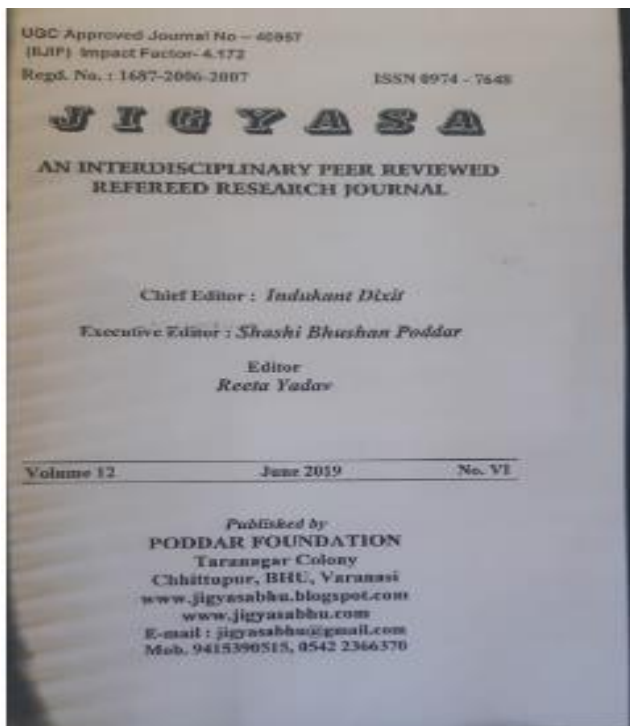
The principle of geochemical existence of F^- with Fe^{2+} and Al^{3+} has been used for its removal from water by using the metal ions loaded active hydrogels. Hydrogel based on the modified acrylic acid (AAc) has been prepared for use in fluoride removal from aqueous systems. AAc was reacted with dimethyl amine to obtain N, N-Dimethylacrylamide (DMAAm) that was subsequently polymerized using ammonium persulphate (APS) as initiator. The resulting poly(N, N-Dimethylacrylamide) (PDMAAm) was further copolymerized with AAc as the second component in the presence of N,N-methylenebisacrylamide (MBAAm) to get a crosslinked

bifunctional hydrogel poly(N,N-DMAAm-co-AAc) [P(DMAAm-co-AAc) hydrogel] that has both amide and acid functional groups. For comparison, reference poly(acrylic acid) (PAAc hydrogel) was prepared by first polymerizing AAc with APS followed by crosslinking with MBAAm using AAc in equal weight, as the second component. The PDMAAm was characterized by nitrogen analysis, FTIR and NMR; while its hydrogel with AAc and reference PAAc hydrogel were characterized by nitrogen analysis, FTIR, SEM, and studying water uptake behavior, as a function of time, temperature, pH and in 5 % NaCl solution. The hydrogels were separately loaded with Fe^{2+} and Al^{3+} , and were used as adsorbent for the uptake of F^- as a function of external environmental factors such as temperature and pH.

138

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Interactions of Drug Doxycycline Hyclate with Galactitol in Aqueous Solutions at Different Temperatures by Volumetric and Acoustic Methods

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Abstract: The experimental values of density, ρ and speed of sound, u of doxycycline hyclate drug (0.002–0.014) mol kg⁻¹ in water and (0.1, 0.2 and 0.4) mol kg⁻¹ of aqueous galactitol solutions at temperatures $T = (303.15, 308.15 \text{ and } 313.15) \text{ K}$ and at atmospheric pressure have been reported in the present communication. From the experimental values, various derived parameters such as apparent molar volume (Φ_V), apparent molar isentropic compression (Φ_K), limiting apparent molar volume (Φ_V^0), limiting apparent molar isentropic compression (Φ_K^0), limiting apparent molar volume of transfer ($\Delta\Phi_V^0$), limiting apparent molar isentropic compression of transfer ($\Delta\Phi_K^0$), limiting apparent molar expansibility (Φ_E^0), thermal expansion coefficient (α) and acoustic parameters like isentropic compressibility (κ_S), intermolecular free length (L_f), and specific acoustic impedance (Z) were calculated. The structure-making behaviour of DH in aqueous galacti-

tol solution was determined on the basis of Hepler's Equation i.e. on the basis of sign of $\left(\frac{d^2\Phi_V^0}{dT^2}\right)_P$. The various derived parameters were utilised to interpret the molecular interactions i.e. solute–solute and solute–solvent existing in the studied system.


Keywords: apparent molar properties; doxycycline hyclate; galactitol; molecular interactions.

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Thermophysical properties of glycine and glycyglycine in aqueous tartaric acid at different temperatures: Volumetric, acoustic and viscometric studies

Shashi Kant Sharma  , Abhishek Thakur, Dinesh Kumar, Vikas Nathan

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Highlights

- Molecular interactions between glycine and glycyglycine in water and in aqueous tartaric acid solutions are studied.
- The Volumetric, Compressibility, Acoustic and Viscometric methods are used to interpret results.
- Solute-Solvent interactions predominate over solute-solute interactions.
- Glycine and glycyglycine behaves as a structure breaker in water and in aqueous tartaric acid solutions.

Abstract

Thermophysical properties like density, **ultrasonic velocity** and viscosity of glycine, **glycyglycine** in water and in (0.1942, 0.3773, 0.5504 and 0.7142) mol·kg⁻¹ aqueous **tartaric acid** solutions as a function of concentration at different temperatures ranging between (298.15 and 318.15) K have been determined. These data have been utilized to calculate apparent **molar volume** (φ_v), apparent molar isentropic **compressibility** (φ_k), and viscosity B-coefficient values of the studied solutions. The **partial molar volumes** (φ_v^0), partial molar isentropic compressibility (φ_k^0) and experimental slopes (S_v^* and S_k^*) derived from the Mason equations have been interpreted in terms of solute-solute and solute-solvent interactions. The viscosity coefficients A and B have been determined from the Jones-Dole equation. From the volumetric and viscometric data, **hydration number** (n_H) has been calculated and further, the structural effects of glycine, glycyglycine in tartaric acid solution has been discussed. The results were explained in terms of structure making and structure breaking properties.



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Structural, magnetic and Mössbauer analysis of lanthanum and nickel doped Co₂Y-type hexaferrite nanomaterial matrix synthesized by sol-gel auto-combustion technique

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ABSTRACT

Sol-gel auto-combustion technique was used to prepare the lanthanum (La³⁺) and nickel (Ni²⁺) doped Co₂Y-type Sr_{1-2y}La_yCo₂Ni_xFe_{12-x}O₂₂; x = 0.0, 0.1, 0.2, 0.3; y = 0.0, 0.01, 0.02, 0.03 nanohexaferrites materials. Synthesized nanohexaferrites were characterized by XRD, FTIR and EDS analysis. The collective XRD analysis reveals the single-phase patterns corresponding to various dopants concentration in the powder system matrix. The calculated structural - auxiliary parameters viz. Lattice constants (a and c), cell volume (V_{cell}), X-ray density (d_x), bulk density (d_b) and porosity (P) have been found in range; a = 5.8798–5.8321 Å, c = 43.7316–44.7214 Å, V_{cell} = 1309.303–1317.304 Å³, d_x = 5.014–4.987 gm/cm³, d_b = 3.774–3.128 gm/cm³ and P = 0.247–0.37. The XRD results were supported by FESEM and TEM images, which showed the plate-like shape of particles. Unlike irregular grains, regular platelet-like grains were composed of several crystallites with almost parallel c-axis and this is the most suitable shape for microwave absorption. Fourier-transform infrared spectroscopy (FTIR) showed the location of the ions including their respective bonds within the structure of the lattice matrix and variation of ion concentration was depicted by EDS analysis of the powder matrix. The investigation of magnetic properties was done by vibrating sample magnetometer (VSM). The saturation magnetization (M_s) and coercivity (H_c) were found to be in the range of 34.29–43.711 emu/g and 16.27–299.441 Oe, respectively. The improved values of M_s and H_c mean that these powders can be used as pre-eminent contestant materials for perpendicular recording media (PMR) applications. In addition, the synthesized Co₂Y-type nanohexaferrites were characterized for the Mössbauer analysis at room temperature. The normal Zeeman (six-line patterns) splitting confirms the hexagonal phase of synthesized nanohexaferrites.

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1. Introduction

Ferrites of magneto plumbite crystal structure have a wide scope of utilization in the field of materials science [1]. Especially,

strontium hexaferrites were extensively considered due to their unique electromagnetic properties, good thermal durability, corrosion resistivity, perfect chemical stability. Strontium hexaferrites have broad high-frequency applications in various electromagnetic gadgets from microwave to radio frequencies due to their high electrical resistivity, also applicable in transformer core including high-quality filters [2,3]. Due to the integration of more components in smaller space, the compatibility of electromagnetic interference (EMI), as well as anti-electromagnetic interference, become one of the most thought-provoking challenges in the high-frequency microelectronic products. Co₂Y-type hexagonal ferrites

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Fabrication of Ni²⁺ and Dy³⁺ substituted Y-Type nanohexaferrites: A study of structural and magnetic properties

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ABSTRACT

A series of the Y-type hexaferrites having chemical composition Sr₂₋₄Dy_xCo₂Ni_yFe_{12-3y}O₂₂ (x = 0.0, 0.01, 0.02, 0.03, 0.04 and y = 0.0, 0.1, 0.2, 0.3, 0.4) were fabricated with doping of Dy³⁺ and Ni²⁺. XRD patterns confirm the formation of a single-phase hexagonal structure along with some impurity phases. FESEM and TEM analysis shows the formation of agglomerated magnetic nanoparticles. M – H loops obtained from VSM observations were used to study the magnetic parameters viz. saturation magnetization (M_s = 34.28–43.30 emu/g), remanence (M_r = 1.08–11.60 emu/g) and coercivity (H_c = 33.98–246.05 O_e), which were found to increase with an increase in Dy³⁺/Ni²⁺ substituents. The synthesized materials with excellent structural and magnetic properties are promising candidates for high-frequency applications.

1. Introduction

The nanocrystalline hexagonal ferrites have gained a lot of scientific attention and researchers are working on Y-type hexagonal ferrites due to their high chemical stability, unique magnetic properties and low cost [1–4]. Today, the world is demanding technology with high performance electronic and telecommunication devices. This demand can be fulfilled with the help of Y-type hexaferrites because these materials can operate at high frequency. A high frequency, high permeability, and planar magnetocrystalline anisotropy enhance the practical applicability of such magnetic materials.

The Y-type hexaferrites have hexagonal crystal structure with ferromagnetic nature and the space group is R3m; the molecular unit consists of two and four-layered blocks termed as S and T blocks respectively [5]. The technology demands the materials with high numbers of functionalities and a high-frequency range for useful applications [6]. These hexaferrites have general formula A₂Me₂Fe₁₂O₂₂, where A and Me refers to divalent metal ions distributed among six (two tetrahedral and four octahedral) sublattices [7]. Y-type hexagonal ferrites have wide applications in high frequency operating devices and

high-quality filters due to their high chemical stability and unique magnetic properties. Also, a high value of coercivity makes them more suitable in microwave devices.

Many techniques like sol-gel, hydrothermal, co-precipitation and glass crystallization are developed to fabricate Y-type hexaferrite [8]. In this paper, the sol-gel method was used for the preparation of nanohexaferrites because this technique can easily control the size, shape and distribution of the particles at low annealing temperatures [9]. Moreover, the co-doping of transition and rare earth metals in Y-type hexaferrites is rarely reported to study the influence on the structural and magnetic properties of hexaferrites. Rare earth materials are known to possess good electrical insulation with high electrical resistivity. Nickel is a corrosion-resistant transition metal and ferromagnetic at room temperature. So, this work aims to study the effect of Dy³⁺ and Ni²⁺ substitutions on structural and magnetic properties of synthesized nanohexaferrites. Strontium based Y-type hexaferrites were substituted with Dy³⁺ ions at Sr²⁺ and Ni²⁺ ions at Fe³⁺ ions and magnetic properties were expected to be enhanced.

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Structural, magnetic and Mössbauer analysis of lanthanum and nickel doped Co₂Y-type hexaferrite nanomaterial matrix synthesized by sol-gel auto-combustion technique

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ABSTRACT

Sol-gel auto-combustion technique was used to prepare the lanthanum (La³⁺) and nickel (Ni²⁺) doped Co₂Y-type Sr_{1-2x}La_yCo₂Ni_xFe_{12-2x}O₂₂; x = 0.0, 0.1, 0.2, 0.3; y = 0.0, 0.01, 0.02, 0.03 nanohexaferrites materials. Synthesized nanohexaferrites were characterized by XRD, FTIR and EDS analysis. The collective XRD analysis reveals the single-phase patterns corresponding to various dopants concentration in the powder system matrix. The calculated structural - auxiliary parameters viz. Lattice constants (a and c), cell volume (V_{cell}), X-ray density (d_x), bulk density (d_b) and porosity (P) have been found in range; a = 5.8798–5.8321 Å, c = 43.7316–44.7214 Å, V_{cell} = 1309.303–1317.304 Å³, d_x = 5.014–4.987 gm/cm³, d_b = 3.774–3.128 gm/cm³ and P = 0.247–0.37. The XRD results were supported by FESEM and TEM images, which showed the plate-like shape of particles. Unlike irregular grains, regular platelet-like grains were composed of several crystallites with almost parallel c-axis and this is the most suitable shape for microwave absorption. Fourier-transform infrared spectroscopy (FTIR) showed the location of the ions including their respective bonds within the structure of the lattice matrix and variation of ion concentration was depicted by EDS analysis of the powder matrix. The investigation of magnetic properties was done by vibrating sample magnetometer (VSM). The saturation magnetization (M_s) and coercivity (H_c) were found to be in the range of 34.29–43.711 emu/g and 16.27–299.441 Oe, respectively. The improved values of M_s and H_c mean that these powders can be used as pre-eminent contestant materials for perpendicular recording media (PMR) applications. In addition, the synthesized Co₂Y-type nanohexaferrites were characterized for the Mössbauer analysis at room temperature. The normal Zeeman (six-line patterns) splitting confirms the hexagonal phase of synthesized nanohexaferrites.

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1. Introduction

Ferrites of magneto plumbite crystal structure have a wide scope of utilization in the field of materials science [1]. Especially,

strontium hexaferrites were extensively considered due to their unique electromagnetic properties, good thermal durability, corrosion resistivity, perfect chemical stability. Strontium hexaferrites have broad high-frequency applications in various electromagnetic gadgets from microwave to radio frequencies due to their high electrical resistivity, also applicable in transformer core including high-quality filters [2,3]. Due to the integration of more components in smaller space, the compatibility of electromagnetic interference (EMI), as well as anti-electromagnetic interference, become one of the most thought-provoking challenges in the high-frequency microelectronic products. Co₂Y-type hexagonal ferrites

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Fabrication of Ni²⁺ and Dy³⁺ substituted Y-Type nanohexaferrites: A study of structural and magnetic properties

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ABSTRACT

A series of the Y-type hexaferrites having chemical composition Sr_{2-x}Dy_xCo₂Ni_yFe_{12-y}O₂₂ (x = 0.0, 0.01, 0.02, 0.03, 0.04 and y = 0.0, 0.1, 0.2, 0.3, 0.4) were fabricated with doping of Dy³⁺ and Ni²⁺. XRD patterns confirm the formation of a single-phase hexagonal structure along with some impurity phases. FESEM and TEM analysis shows the formation of agglomerated magnetic nanoparticles. M – H loops obtained from VSM observations were used to study the magnetic parameters viz. saturation magnetization (M_s = 34.28–43.30 emu/g), remanence (M_r = 1.08–11.60 emu/g) and coercivity (H_c = 33.98–246.05 O_e), which were found to increase with an increase in Dy³⁺/Ni²⁺ substituents. The synthesized materials with excellent structural and magnetic properties are promising candidates for high-frequency applications.

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A study of magnetic properties of Y–Ni–Mn substituted Co₂Z-type nanohexaferrites via vibrating sample magnetometry

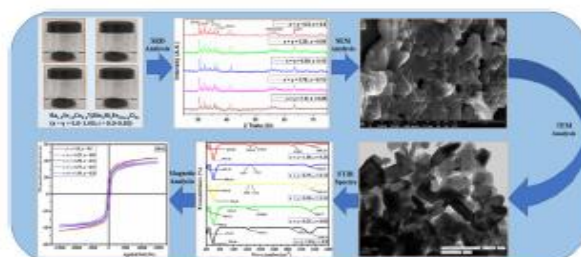
Kirti Singha¹ · Rohit Jasrotia¹ · Virender Pratap Singh^{1,2} · Monika Chandel¹ · Rajesh Kumar¹ · Susheel Kalia³

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Abstract

Z-type nanohexaferrites has attracted substantial attention in the field of materials science especially in the research areas of high frequency and microwave devices applications due to their excellent magnetic properties in comparison with the spinel ferrites. A series of Ba_{1.5}Sr_{1.5}Co_{2-x}Y_xMn_yNi_zFe_{24-y-z}O₄₁ ($x = 0.0, 0.05, 0.10, 0.15, 0.20$; $y = z = 0.0, 0.25, 0.50, 0.75, 1.00$) Z-type hexaferrites were fabricated with the help of sol-gel auto-combustion method in order to achieve its better microstructural and magnetic properties. The crystallographic structure of prepared specimens was characterized by XRD which revealed the presence of Z-type single-phase hexagonal structure with some additional phases accompanied by P6₃/mmc space group. The average particle size calculated by Scherrer formula was found to be in the range of 40–55 nm. The FTIR spectrum of synthesized specimens was observed in the range of 400–600 cm⁻¹ which confirms our microstructural results to a great extent. From the *M*–*H* hysteresis loops, magnetic parameters such as saturation magnetization (*M_s*) (44.04–35.59 emu/g) and coercivity (*H_c*) (42.3–248.96 Oe) were measured and it was observed that the synthesized specimens exhibit excellent characteristics of soft ferrite which make it suitable for the multilayer chip inductors, electromagnets, and magnetic storage devices applications.

Graphical Abstract



Keywords Z-type nanohexaferrites · Sol-gel preparation · Structural study · Morphological Investigation · Magnetic measurements

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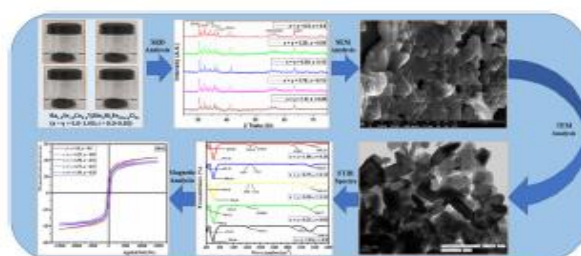
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A NOTE ON VARIANCE BOUNDS AND LOCATION OF EIGENVALUES

R. SHARMA, A. SHARMA AND R. SAINI

(Communicated by M. Krnić)

Abstract. We discuss some extensions and refinements of the variance bounds for both real and complex numbers. The related bounds for the eigenvalues and spread of a matrix are also derived here.

1. Introduction

Let z_1, z_2, \dots, z_n denote n complex numbers. Their arithmetic mean is the number

$$\frac{1}{n} \sum_{i=1}^n z_i = \tilde{z}. \tag{1.1}$$

In literature, the number

$$\frac{1}{n} \sum_{i=1}^n |z_i - \tilde{z}|^2 = S_z^2 \tag{1.2}$$

or its equivalent expressions have been studied in several different contexts and notations and is termed as the variance of complex numbers at many places. For example, see Audenaert [2], Bhatia and Sharma [4,5], Merikoski and Kumar [13], and Park [17].

The number

$$\frac{1}{n} \sum_{i=1}^n (z_i - \tilde{z})^2 = S^2 \tag{1.3}$$

is also important in this context. If z_i 's are all real we denote them by x_i 's with $a = \min x_i$ and $b = \max x_i$. The arithmetic mean by \bar{x} and variance by the lower case letter s^2 . In this case $S_z = |S| = S = s$ but in general S_z rather than $|S|$ is more consistent with s . For instance, $s = 0$ ($S_z = 0$) if and only if all the x_i 's (z_i 's) are equal. This is not the case with $|S|$; for example, for three distinct complex numbers $0, \pm \frac{1}{2} + i\frac{\sqrt{3}}{2}$ we have $S = 0$. It however turns out that for some purposes s^2 is more consistent with

$$\sigma_z^2 = \frac{|S^2| + S_z^2}{2} \tag{1.4}$$

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RESEARCH ARTICLE

ROLE OF AGRICULTURE SECTOR IN THE ECONOMIC DEVELOPMENT OF HIMACHAL PRADESH

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ABSTRACT

Agriculture sector in India is of such relevance that it is contributing employment to more than 90 percent of the work force and 17 to 18 percent of the Nation's Gross Domestic Product. It is pertinent to understand that the land available for cultivation is abundant and in a country like India where large majority of the population belongs to middle and low economic class and youth is in search of job avenues, if youth is skillfully involved to initiate and go for adopting innovative agricultural activities, employability can be enhanced. At present, land available for cultivation out of the total land holdings in Himachal Pradesh is 71.7 percent of the total area and 87.9 percent of sown area in India. Further, Net Area sown is 11.8 percent of the total cultivable area in the State as compared to 45.8 percent that of the Country. Agriculture sector is contributing nearly 16 percent to the Gross State Domestic Product of Himachal Pradesh. Based upon the criteria of land holding, suitability, cultivable crropped area, net sown area and irrigation, the research paper analyses to examine the role of agriculture sector in the economic development of Himachal Pradesh.

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INTRODUCTION

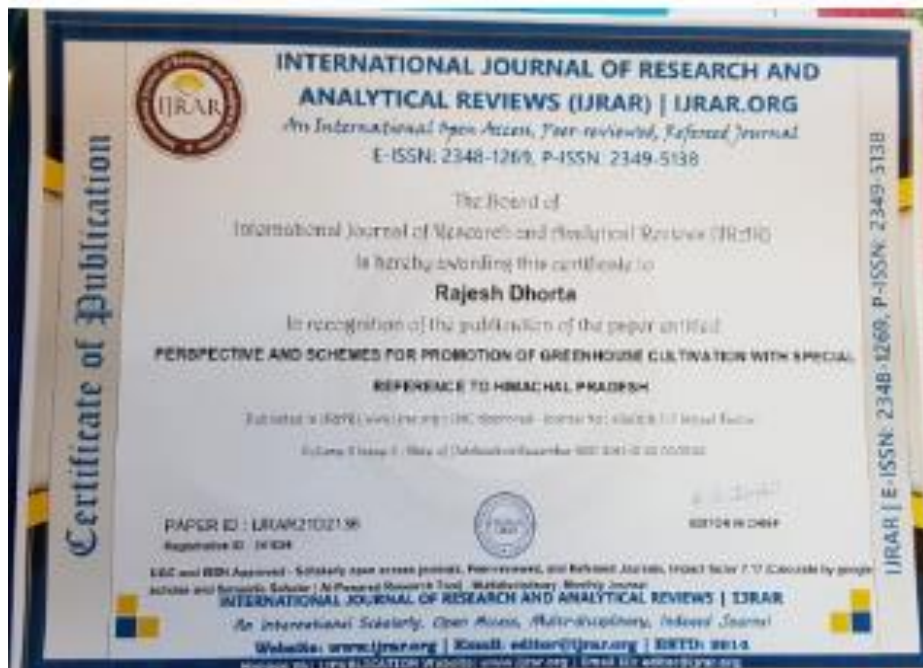
India has diversity of situation and land use characteristics. The Country is spread over the total land area of 329 million hectares out of which, gross cropped area is 141 million hectares. The agriculture of the Country is of such relevance that it is contributing employment to more than 90 percent of the work force and 17 to 18 percent of the Nation's Gross Domestic Product. It is pertinent to understand that the land available for cultivation is abundant and in a country like India where large majority of the population belongs to middle and low economic class and youth is in search of job avenues, if youth is skillfully involved to initiate and go for adopting innovative agricultural activities, employability can be enhanced. At present, land available for cultivation out of the total land holdings in Himachal Pradesh is 71.7 percent of the total area and 87.9 percent of sown area in India. Further, Net Area sown is 11.8 percent of the total cultivable area in the State as compared to 45.8 percent that of the Country. It is further pertinent to mention that this sector has good chances of yield and earning as it is being facilitated by resources.

In Himachal Pradesh, 20 percent of the net cropped area is equipped with the irrigation facilities and in India, 45 percent of the net cropped area is having irrigation facility. Having such a relevant sector of the national economy, agriculture sector needs to be prioritized in terms of recognizing the scope of technology. The research paper analyses to examine the role of agriculture sector in economic development of Himachal Pradesh.

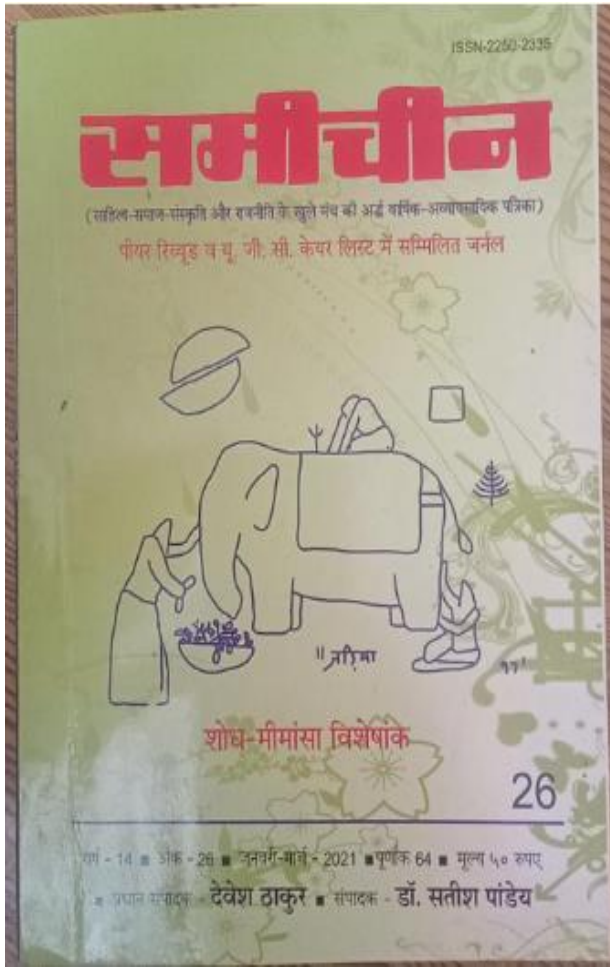
REVIEW OF LITERATURE

Cassallary (2016) has claimed that majority of marginal and small farmers' practice in traditional farming in the state of Himachal is not remunerative. He feels that a manifold increase in the resource-use efficiency in crop production can be obtained through protected cultivation compared to open-field conditions which can help marginal and small farmers by providing them financial and extension services for infrastructural development and transfer of technology speeded up to bring the desired technology to the region.

Dr. Rajesh Kumar Dhorta



Dr. Satya Narain Snehi



क्र. सं.	विषय	पृष्ठ
1.	अपने सौं	06
2.	बदलते सांस्कृतिक संदर्भों की तलाश और 'बसंत पहाड़' - डॉ. अनिल सिंह	7-12
3.	समकालीन कविता में ग्रामीण संवेदना और लोककल्याण के कवि कुमार कृष्ण - डॉ. सत्यनारायण स्नेही	13-20
4.	वैश्विकरण के परिदृश्य में हिंदी कविता - डॉ. सचिन पट्ट	21-24
5.	'सच तो यही है' : जगत कवयित्रीयों का प्राथमिक दस्तावेज - डॉ. प्रवीण चंद मिश्र	25-32
6.	रचना-विधान में संकेत और व्यंग्य की अर्थवत्ता - डॉ. प्रदीप शिखरी	33-40
7.	लीलाधर जगृदी की कविता में विशिष्टता बोध - प्रदीप सिंह	41-47
8.	विनेट रस्तोमी कृत्र 'जर्क की भीमार' को रोमंचनीयता - डॉ. दुर्गादेवी शीतल मुल्लाबराव	48-53
9.	साहित्य और सिनेमा का अंतः संबंध - डॉ. शेकुल मोरछ शीरसागर	54-59
10.	जनसंघार के माध्यमों का लोक-साहित्य व लोक-संस्कृति के संघर्ष में योगदान - डॉ. गजेन्द्र भारद्वाज	60-66
11.	पाश्चुरातक लेखन में बामू श्यामसुन्दर दास का प्रदेश और प्रासंगिकता - डॉ. गीता शर्मा	67-73
12.	जनसंघार-जगत का कारणात्मक अनुशीलन - प्रा. अशोक गुडे/डॉ. भगवती प्रसाद उपाध्याय	76-80
13.	विमर्श, विचारधारा और दृष्टान्त - अनन्त द्विवेदी	81-88
14.	डॉ. उमेश चंद्र शुक्ल की गजालों में अधिष्ठात आत्म आदमी की पीड़ा - डॉ. चंद्रभूषण शुक्ल	89-96
15.	स्त्री आत्मकथाओं में सत्ता संबंध (मैत्रेयी पुष्पा और तसलीमा नसरिन के विशेष सन्दर्भ में) - शिव दात	97-102
16.	सुधा ओम खिंगर के कहानी संग्रह 'कौन सी जमीं अपनी' की मूल संवेदना - डॉ. पंडार राही खान, डॉ. निरंज	103-109
17.	हिंदी सिन्धु संगीत में जलज - एक विशेषण - डॉ. मोरच जैन	110-113

समकालीन कविता में ग्रामीण संवेदना और लोककल्याण के कवि कुमार कृष्ण

- डॉ. सत्यनारायण स्नेही

जिस दुनिया में लिखी जाएगी बेहतर कविताएँ वही होगी बेहतर दुनिया शब्द और अर्थ नहीं है कविता सबसे सुन्दर सपना है सबसे अच्छे आदमी का

समकालीन हिन्दी कविता में सत्यनारायण प्रसाद तिवारी की कविता का यह कवितांश जिस बेहतर दुनिया की बात करता है। इसी परम्परा के अगले कवि हैं कुमार कृष्ण, जो वैश्विक परिवर्तन और तकनीकी युग के इस दौर में ग्रामीण जीवन की वास्तविकताओं को सहज रूप में व्यक्त करते हुए मानवीय मूल्यों एवं संवेदनाओं की अहमियत बताते हैं और आज के समय में अपनी संस्कृति, सभ्यता, भाषा, एवं अस्मिता बचाने की बात करता है-

मनुष्य के पास होनी चाहिए / बच्चों की शरारतें
बुजुर्गों की यादों का बटुआ / प्यार और नफरत के
रंग-बिरंगे छत्र / उसके पास होने चाहिए / बेशुमार दरवाजे
जो खुल जायें सिर्फ एक बार खटखटाने पर।

कवि कुमार कृष्ण की कविताओं में भारतीय किसान की चिन्ता, व्यथा और पीड़ा विद्यमान है। इन्होंने पहाड़ी अंचल के पारों, समस्याओं, सम्बन्धों, प्राकृतिक एवं सामाजिक परिवेश के समग्र रूपों, परम्पराओं अर्थात् अंचल के समग्र जीवन को अपनी कविता में सरल एवं स्वाभाविक रूप से व्यक्त किया है इनकी कविताओं में भारतीय गाँव, भारतीय समाज तथा परिवार, भारतीय संस्कृति तथा परम्पराएँ, मानवीय रिश्तों के अहसास को देखा जा सकता है। इनमें लोक चेतना तथा ग्रामीण संवेदना का पारदर्शी समन्वय देखने को मिलता है। इनकी कविताओं में पहाड़ी गाँव का दर्द है जो हर गाँव की पीड़ा ब्याप्त करता है। कवि कुमार कृष्ण की कविताओं का मूल स्वर है-

मैं लिखना चाहता हूँ / खेतों के ताप / मिट्टी की बौखलाहट
बीज की बेचैनी / जमीन का उन्माद / सभी कुछ एक साथ
फोड़ना चाहता हूँ पहाड़ / कविता के शब्दों से।

कुमार कृष्ण की कविताओं में पहाड़ का गाँव है, यहाँ के किसानों के सुख-दुख हैं और यह एहसास भी है कि आज के परिवर्तनशील और शहर की ओर बढ़ते हुए वर्गीय जीवन में उसकी क्या स्थिति है? वास्तव में, कृषक जीवन की दास्तान ही ऐसी है कि वह जन्म से मृत्यु पर्यन्त संघर्षमय रहते हुए अभावग्रस्त जीवन-बसर करता है। लोक जीवन में सामान्य व्यक्ति की

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